

Multifunctional Nanoparticles for Medical Imaging

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We present our recent activities in the research field of medical diagnostics on the development of multifunctional biocompatible nanoparticles (NPs). Our work is focused on the synthesis and characterization of **calcium fluoride** and **iron oxide** based NPs.

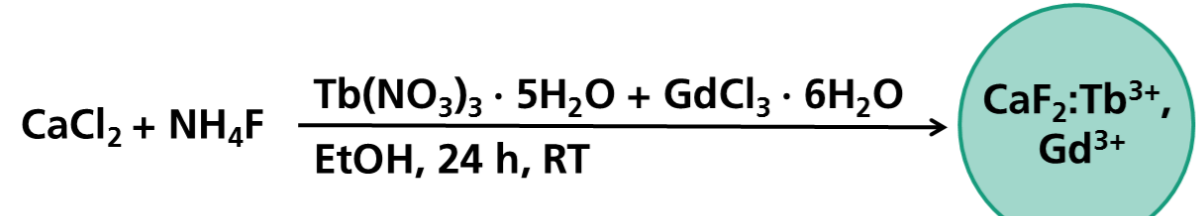
The **CaF₂-based** NPs are produced by wet-chemical synthesis [1] and doped with multiple lanthanide ions, e.g. Gd³⁺ and Tb³⁺, leading to paramagnetism and fluorescence, making them suitable for T₁-weighted magnetic resonance imaging (MRI) and fluorescence microscopy [2]. The characterization is done by conventional methods such as transmission electron microscopy (TEM), X-ray diffraction analysis (XRD) and

photoluminescence (PL) spectroscopy. The capability of these NPs to be used as positive contrast agents for MRI was investigated. In addition, the cytotoxicity of the NPs was tested by a cell viability assay with respect to a later in vivo application.

The **Fe₃O₄-based** NPs are synthesized by methods such as co-precipitation [3], thermal decomposition [4] or sonochemistry [5] and are optionally coated with silica [6]. The latter is envisaged to improve the biocompatibility. The characterization of these NPs is also done by TEM and XRD. Additionally the magnetization curves were measured with a vibrating sample magnetometer (VSM).

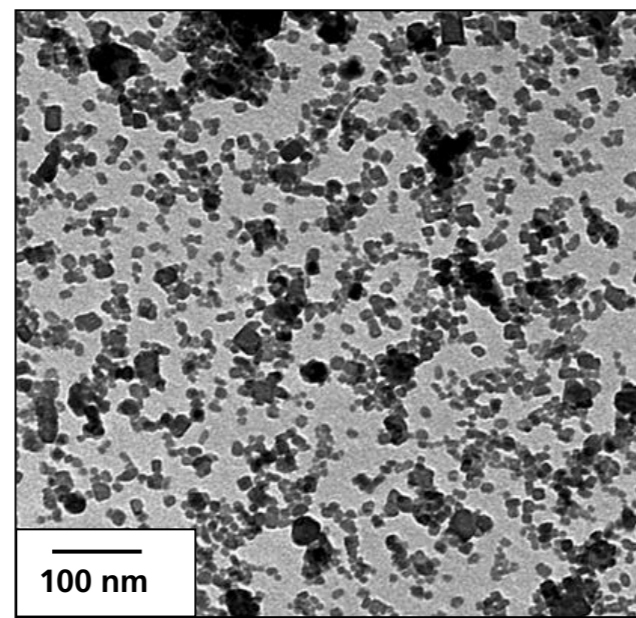
Calcium fluoride

Synthesis and Structural Properties

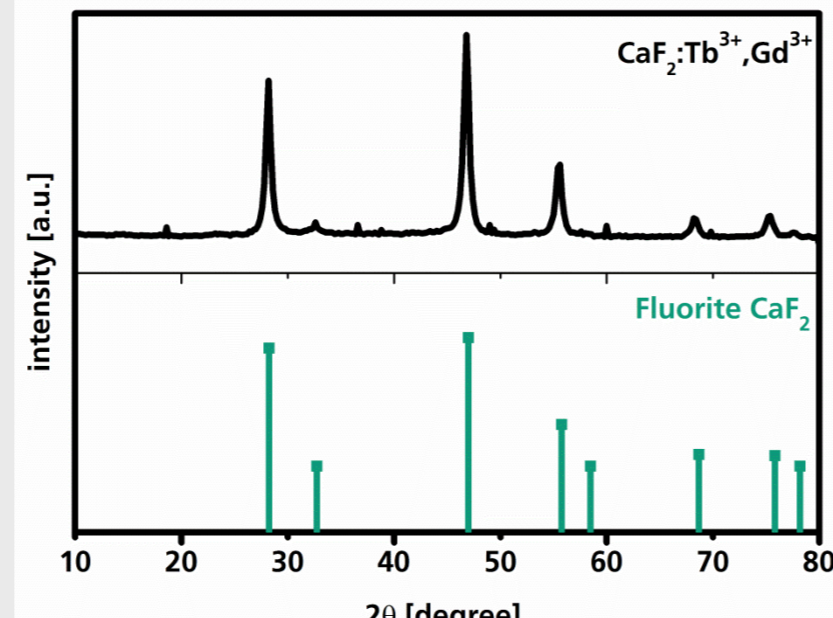


Synthesis of multifunctional NPs on the basis of CaF₂

- Adjustment of NP properties by doping with different lanthanide ions in various concentrations:
 - Gd³⁺ leading to paramagnetism
 - Tb³⁺ or Yb³⁺ and Er³⁺, respectively, inducing fluorescence into the NPs



TEM micrograph of CaF₂:Tb³⁺,Gd³⁺-NPs



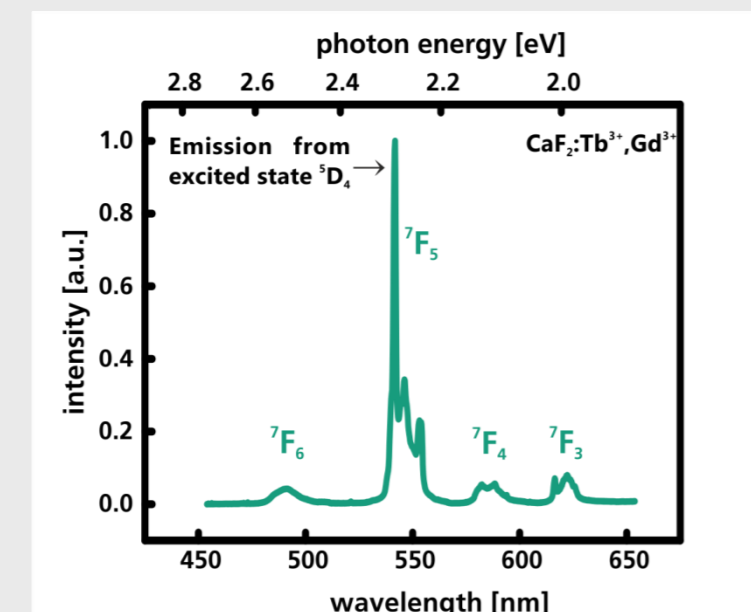
XRD pattern for crystalline CaF₂:Tb³⁺,Gd³⁺-NPs (d = 10 nm, Tb³⁺ and Gd³⁺: 1 mol%)

Optical Properties

- Doping with Gd³⁺ (seven unpaired electrons) leads to paramagnetic CaF₂.



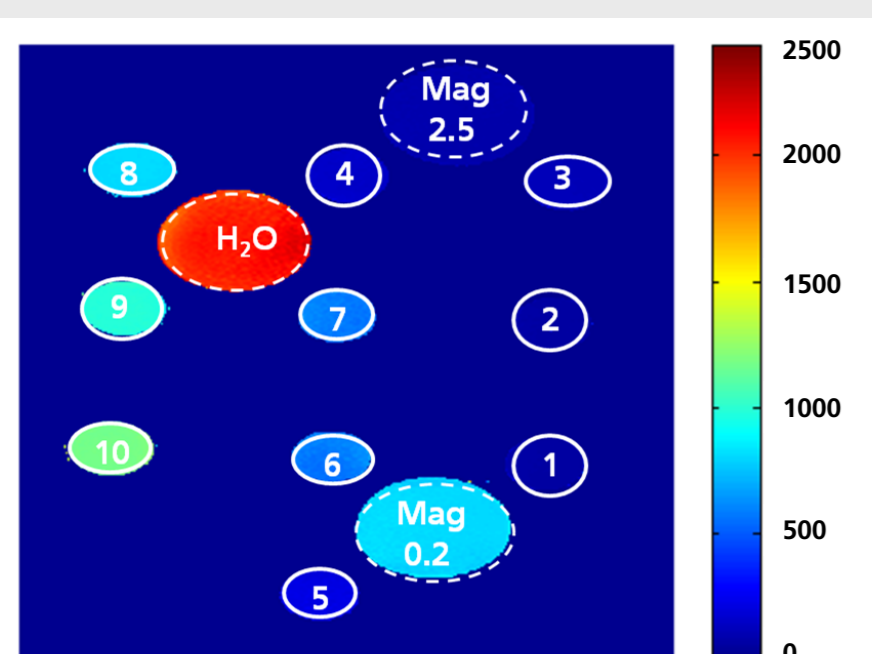
NP-powder of CaF₂:Tb³⁺,Gd³⁺-NPs under excitation with UV-lamp (λ_{ex} = 254 nm)



Normalized PL spectrum of CaF₂:Tb³⁺,Gd³⁺-NPs. The transitions start from the ⁵D₄ excited state of Tb³⁺ and end on the levels indicated (λ_{ex} = 254 nm).

MRI Studies

- Positive contrast agents, such as Gd³⁺ ions induce a shortening of the longitudinal relaxation time T₁ leading to a local signal increase in the MR images.
- The efficiency of the contrast agent is constituted by the relaxivity r₁.



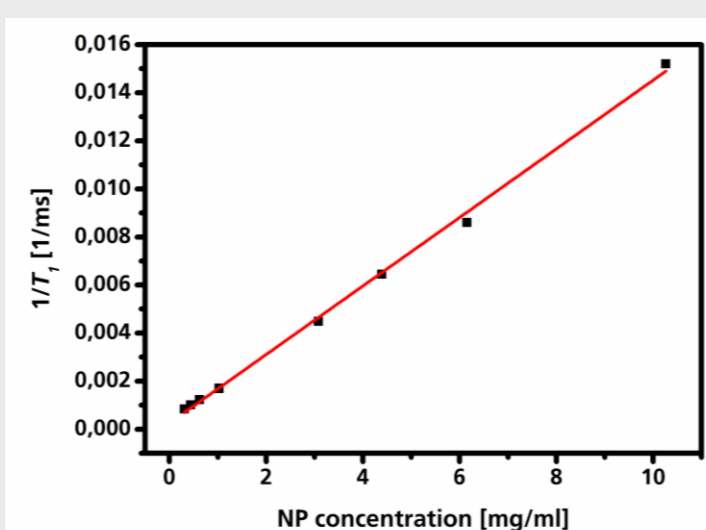
T₁-map from 0 to 2500 ms. (1)-(10): CaF₂:Tb³⁺,Gd³⁺-NP-samples.

References:

Mag2.5 = Magnevist c = 2.5 mmol/l; Mag0.2 = Magnevist c = 0.2 mmol/l and water.

Relaxivity r₁ of CaF₂:Tb³⁺,Gd³⁺-NPs (1 mol% of Gd³⁺): 0.5 ml/mg·s

Relaxivity r₁ of Magnevist: 4.89 ml/mg·s



Inverse relaxation time 1/T₁ of CaF₂:Tb³⁺,Gd³⁺-NP plotted against the concentration

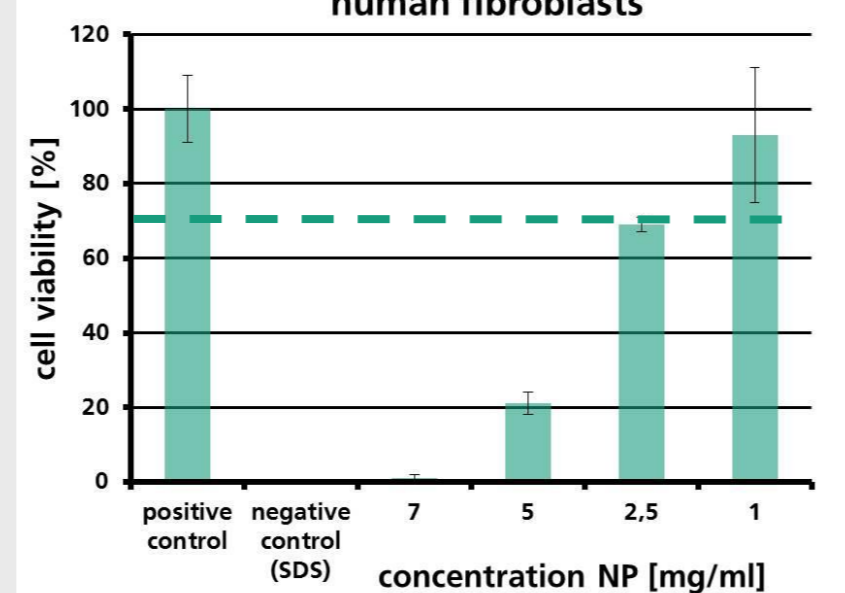
Biocompatibility

- CellTiter-Glo® Luminescent Cell Viability Assay [7] of CaF₂:Tb³⁺,Gd³⁺-NPs on human fibroblasts
- Particle samples with the cellular activity over 70% are biocompatible.

Experimental conditions:

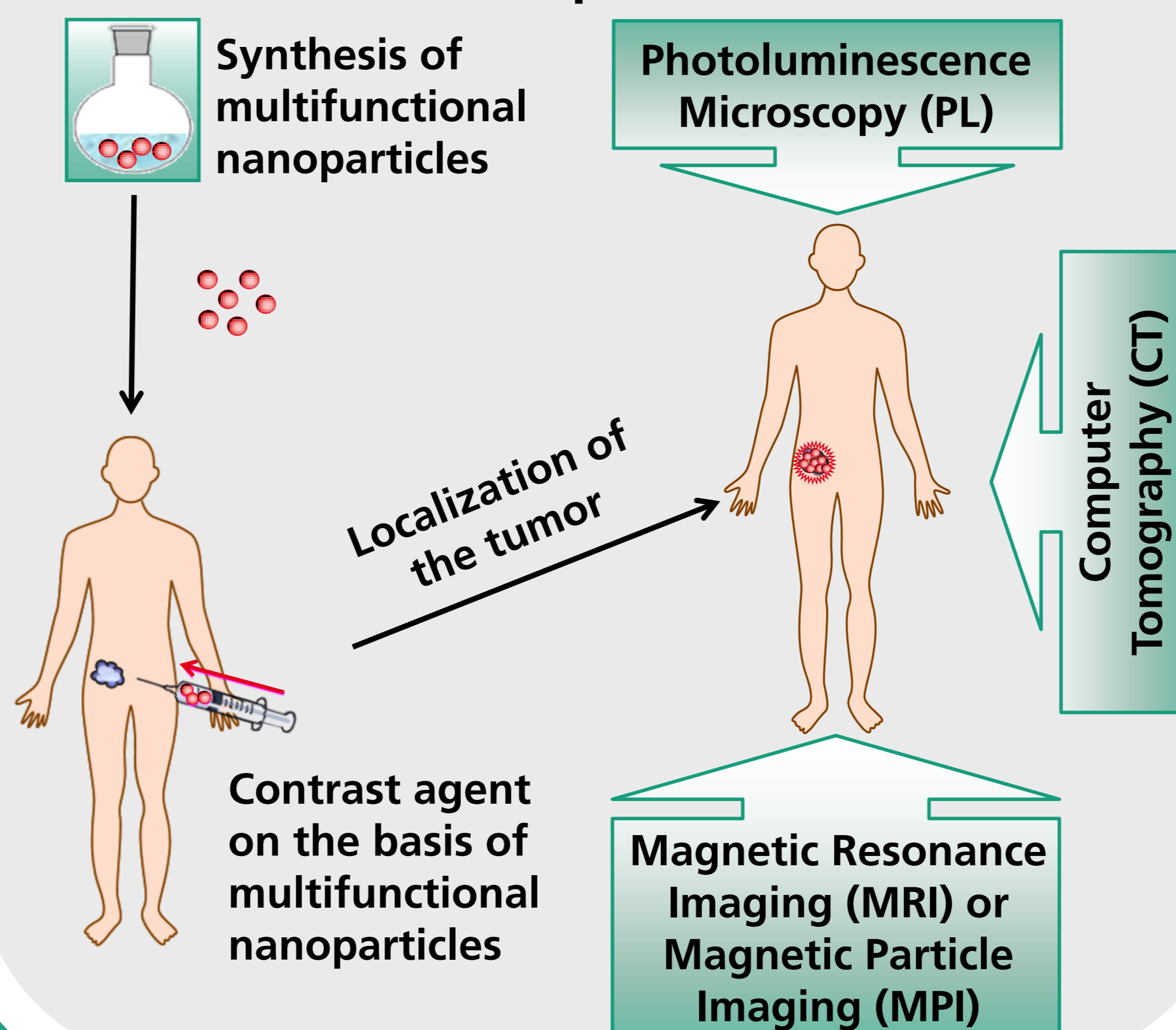
cell culture: human fibroblasts
incubation: 24 h, 37 °C
negative control: Sodium dodecyl sulfate (SDS)
positive control: human fibroblasts in serum-supplemented medium

Cytotoxicity test of CaF₂:Tb³⁺,Gd³⁺-NPs on human fibroblasts

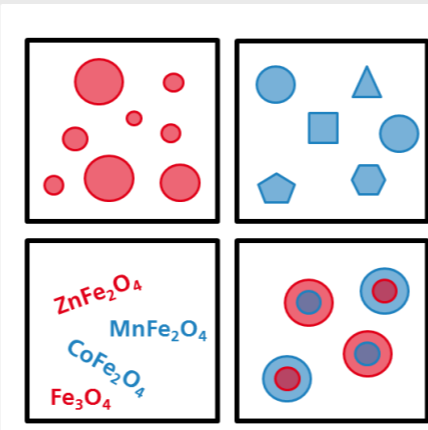


Cell viability [%] for different NP concentrations

Application Potential of Multifunctional Nanoparticles



General idea



Magnetic properties

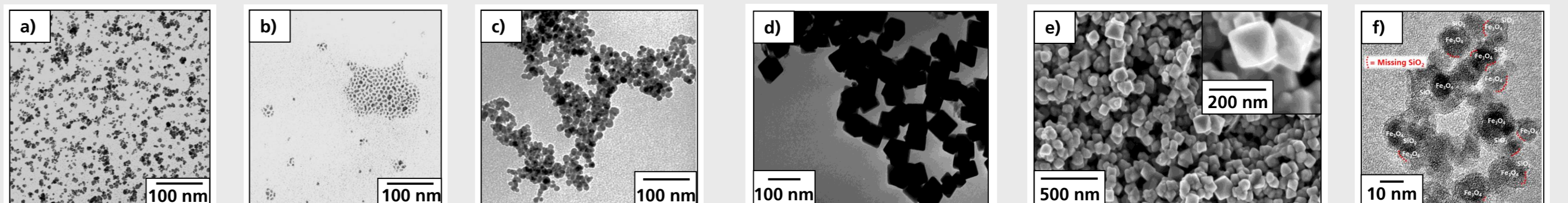
Variation of the morphology, size, composition or structure of the NPs influences the MPI signal

Measured signal in MPI (Magnetic Particle Imaging)

Magnetic properties of Fe₃O₄-based NPs can be adjusted by variation of their:

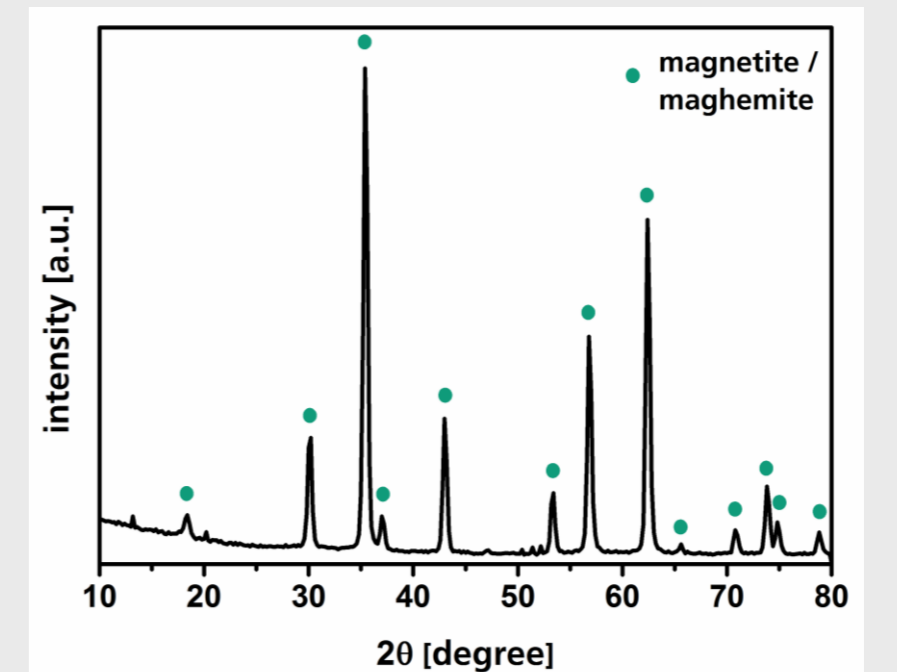
- size
- composition
- morphology or
- architecture.

Synthesis and Structural Properties



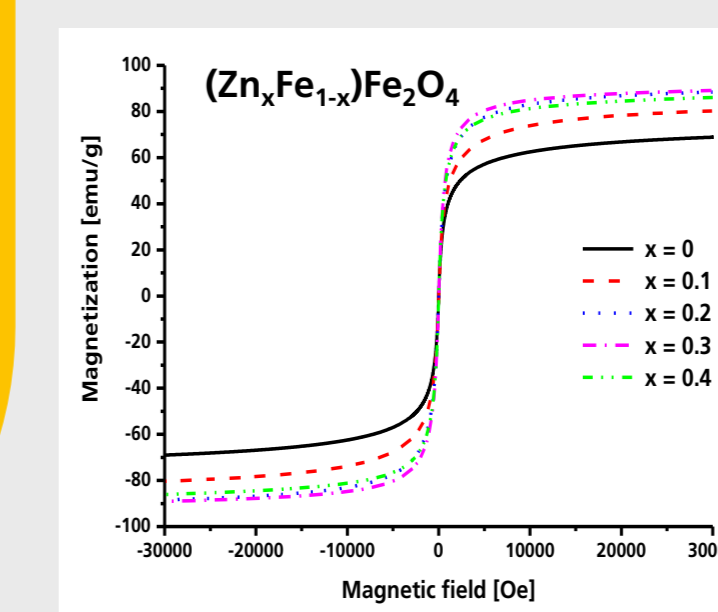
TEM micrographs of Fe₃O₄-NPs after a) co-precipitation, b) thermal decomposition and c) co-precipitation and doping with Zn²⁺. d) TEM image of octahedral NPs obtained after 2d ageing time. e) SEM image of the same sample. The octahedral shape is emphasized in the inset in image e). f) Patchy silica-coated Fe₃O₄-NPs.

Exemplary X-ray diffractogram of Fe₃O₄-NPs. The peaks, indicating the crystallinity of the particles, can either be assigned to magnetite (Fe₃O₄) or maghemite (γ-Fe₂O₃).



Variation of the Composition

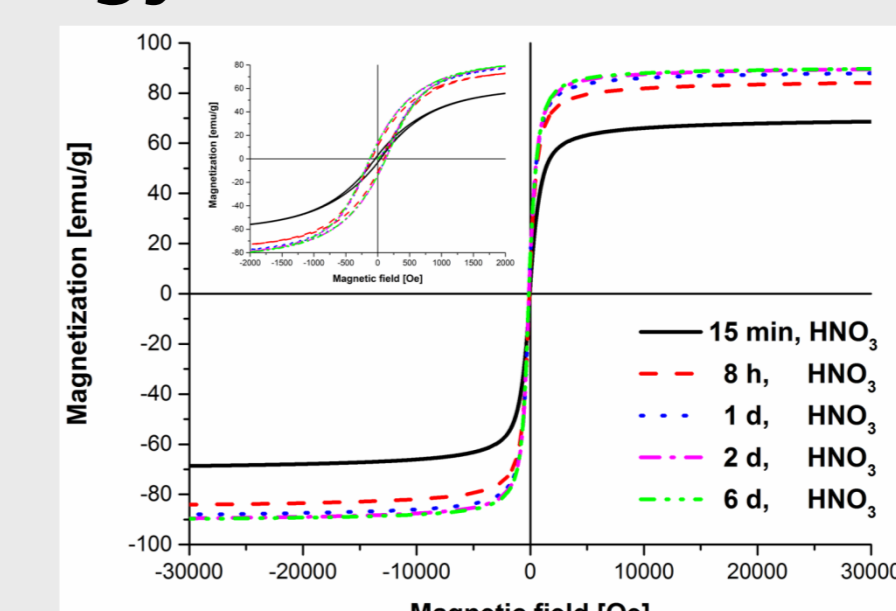
- Doping with Zn²⁺ (TEM c) leads to increased saturation magnetizations. The values vary for different doping concentrations.



Magnetization curves of Fe₃O₄-NPs (doped with x = 0, 0.1, 0.2, 0.3 and 0.4 parts of zinc) after co-precipitation. All samples possess a hysteretic magnetic behavior.

Variation of the Morphology

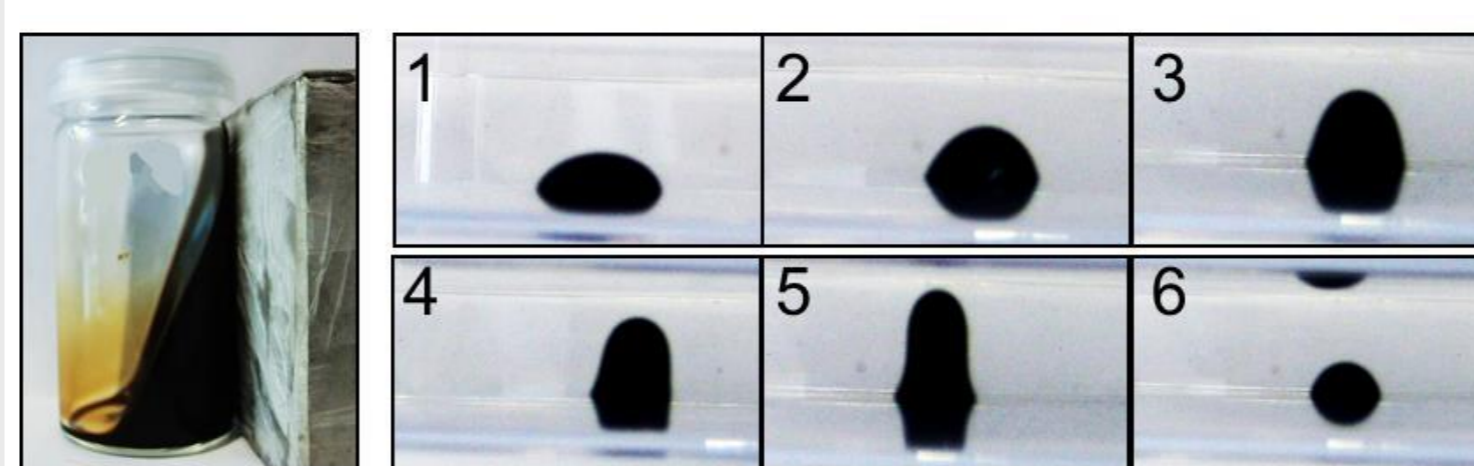
- Combination of ultrasound and nitric acid treatment yields octahedral shape (TEM d,e).



Magnetization curves of Fe₃O₄-NPs after ageing times of 15 min, 8 h, 1 d, 2 d and 6 d. All samples possess a hysteretic magnetic behavior; saturation magnetizations increase with increasing ageing times.

Variation of the Architecture

- A core-shell structure of the NPs (TEM f) can make them biocompatible. Coating with silica results in a true ferrofluid, which is not only stable over a wide range of pH but also in physiological solutions.



Fe₃O₄-NPs coated with silica. Left image: in a vial next to a magnet; right image series: for a single droplet. For the image series 1-6, a magnet was approached from the top.

Summary and Outlook

- Wet-chemical synthesis of multifunctional CaF₂- and Fe₃O₄-based NPs
- Characterization of structural, optical and magnetic properties
- The CaF₂-NPs are biocompatible and suitable for T₁-weighted MRI
- The Fe₃O₄-NPs can be used for MRI and MPI
- Attachment of biomolecules such as antibodies
- Investigation of the X-ray-opacity.

Acknowledgement

Prof. Dr. G. Krohne (Department of Electron Microscopy, Julius-Maximilians-University of Wuerzburg, Germany).

Dr. Marco Metzger (Fraunhofer Institute for Interfacial Engineering and Biotechnology IGB, Project Group Oncology)

Financial Support

This work was financially supported by the Fraunhofer-Gesellschaft zur Foerderung der angewandten Forschung e.V., Munich, Germany.

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